# GENERAL MOTORS CORPORATION SHREVEPORT, LOUISIANA

## EXPANSION ASSESSMENT II PHASE II

**JANUARY, 1991** 

#### PREPARED BY:

C-K ASSOCIATES, INC. 2001 E. 70TH STREET, SUITE 503 SHREVEPORT, LOUISIANA 71105 (318) 797-8636

C-K ASSOCIATES' PROJECT NO. 12-455-1

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#### 1.0 INTRODUCTION

General Motors Corporation's (General Motors) Assembly Plant is a large manufacturing facility located in Shreveport, Louisiana, western Caddo Parish (Figure 1). The plant assembles light duty trucks. The assembly plant occupies approximately 45 acres of a 440 acre site and employs about 2,700 people. Construction of the plant began in 1978 and was completed by 1981. A map of the facility is included as Figure 2 and an aerial photograph is included as Figure 3.

The plant receives its sub-assemblies and parts via railroad freight car and overland truck. The main processes that occur at the plant include: welding of steel sub-assemblies and parts into trucks and sheet-metal assemblies; washing and painting facilities (main hydrocarbon source); trim shop, where inside and outside hardware are assembled to the truck; chassis area, where the engine, axle, transmission and associated parts are assembled to the chassis frame; and the final assembly department that completes assembly operations for a finished truck. Other operations consist of final truck repairs, maintenance, cushion assembly, administrative offices, and other minor associated activities.

Phase II of Expansion Assessment II was initiated when laboratory data indicated the possible presence of 2-Hexanone in the soil at boring NO. 5 (B-5). B-5 is located within the southern portion of the Roll Test Booth Expansion Area (Expansion Area D) as shown on Figure 4. The objective of the Phase II investigation was to confirm the presence of 2-Hexanone in the soil in the vicinity of B-5 and to determine if present, its vertical distribution within the soil.

The findings of the previous assessment at Area D are found in the Expansion Assessment II Report, dated December 1990.

#### 2.0 FIELD INVESTIGATION

Consistent with guidance provided by DEQ (Mr. Leon Waller), four soil borings were drilled within close proximity of B-5 in order to further investigate the possible presence of 2-Hexanone. The borings were placed in locations pre-determined by General Motors' environmental engineers in conjunction with DEQ Groundwater Protection personnel. Each of the four borings was drilled to a depth ranging from 32 to 35 feet below the ground surface. The soil borings are plotted on the Soil Boring Location Map included as Figure 4.

Soil samples were continuously collected with a Shelby tube to the completion depth of each boring. All soil samples and auger cuttings were visually inspected by the onsite hydrogeologist. Detailed boring logs were prepared which included sample numbers, sample depths, visual description of each sample, measured consistency, Unified Soil Classification System (USCS) descriptions, Organic Vapor Analyzer (OVA) measurements, and other pertinent information relative to the drilling operations. The completed soil boring logs are found in Appendix A.

Each soil sample collected was trimmed to remove the smear zone formed during sample acquisition. A portion of each sample was placed within a properly labeled Zip-Lock R bag for OVA analyses. The samples were then disaggregated and allowed to stabilize for approximately 15 minutes prior to scanning with the OVA. The OVA measurements were obtained by inserting the instrument probe into the opening of the Zip-Lock R bag. Following stabilization of the sample head space, the organic vapor concentrations were measured and recorded on the soil boring logs. A second series of OVA measurements were obtained with a light ends filter attached to the instrument probe. The second set of measurements were subtracted from the first set of measurements and the difference recorded on the soil boring logs as the adjusted values. This methodology eliminated the possibility of false high OVA measurements resulting from the presence of natural gas.

Soil samples were collected at five foot intervals for potential laboratory analyses. Samples collected at the shallowest intervals were analyzed first. If 2-Hexanone was detected at concentrations at or above the detection limit in a particular sample, a sample from the interval below it was analyzed. This selection process was continued until concentrations at or above the detection limit failed to be detected.

In order to minimize cross-contamination during sample preparation, each of the samples were handled with a clean pair of surgical gloves and placed on a clean sheet of aluminum foil. All sampling tools were washed with a laboratory grade cleaning compound and deionized water between sample collections.

#### 3.0 EXPANSION AREA D (ROLL TEST BOOTH)

The Roll Test Booth Expansion Area will be used for the purpose of electronically testing assembled vehicles. The foundation for the expansion area will cover an area of 8,100 square feet and will be supported by five reinforced concrete pilings, each approximately 20 feet deep.

On December 10 and 11, 1990, C-K Associates conducted the Phase II assessment at the Roll Test Booth Expansion Area. The subsurface soils were assessed with soil borings B-9 through B-12 to depths of 32 to 35 feet (Figure 4). The soil borings encountered undifferentiated fill to a depth of four feet below the ground surface. Underlying the undifferentiated fill, silty clay was encountered to a depth of 13 to 14 feet. Below the silty clay a homogeneous hard clay with horizontally oriented fine grain sand and silt laminations was encountered. This unit was continuous to the completion depth of both borings. A saturated zone was encountered within each of the borings at a depth of 19 to 20 feet.

Soil samples were collected from each of the borings as described in Section 2.0 and were submitted to West-Paine Laboratories for 2-Hexanone analyses (SW-846, Method 8240). In addition to the submittal of the soil samples two field blanks were also submitted to the laboratory for 2-Hexanone analyses (EPA Method 624).

Field OVA analyses detected organic vapor concentrations ranging from 0 to 25 ppm in soil samples analyzed from one to eight feet below the surface. No measurable readings were recorded from samples collected below the depth of eight feet. The specific values are recorded on the soil boring logs found in Appendix A.

The laboratory analyses indicated the presence of 2-hexanone at B-9 and B-10. At B-9, 2-Hexanone was detected at a concentration of 0.24 mg/kg (Sample No. 901) in a sample collected from a depth of five feet and at a concentration of 0.21 mg/kg (Sample No. 902) in a sample collected from a depth of ten feet. At B-10, 2-Hexanone was detected at concentration of 0.09 mg/kg (Sample No. 1001) at a depth of five feet. The detection limit reported by the laboratory for the 2-Hexanone analyses was 0.05 mg/kg. The laboratory analyses are summarized on Table 1. Completed laboratory reports and chain of custody documentation are found in Appendix B.

Subsequent to attaining the completion depth of each boring, drilling equipment was retracted and the borehole was grouted up to the ground surface with a cement-bentonite slurry.

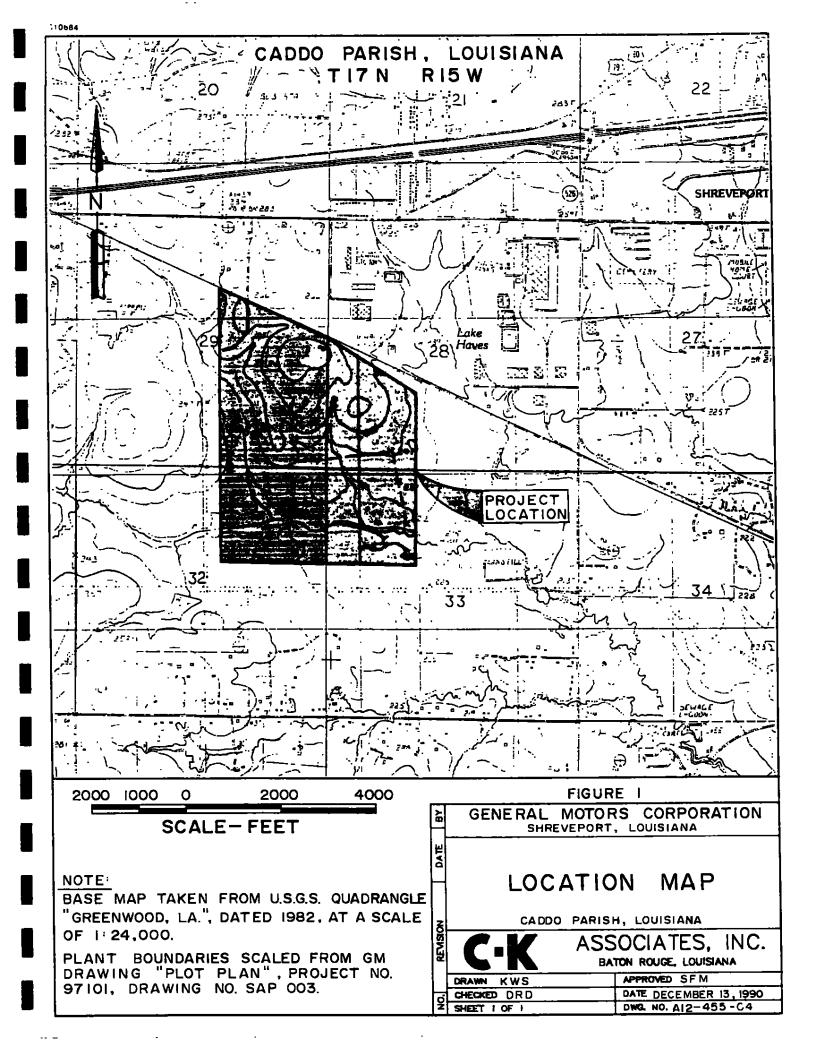
#### 4.0 SUMMARY

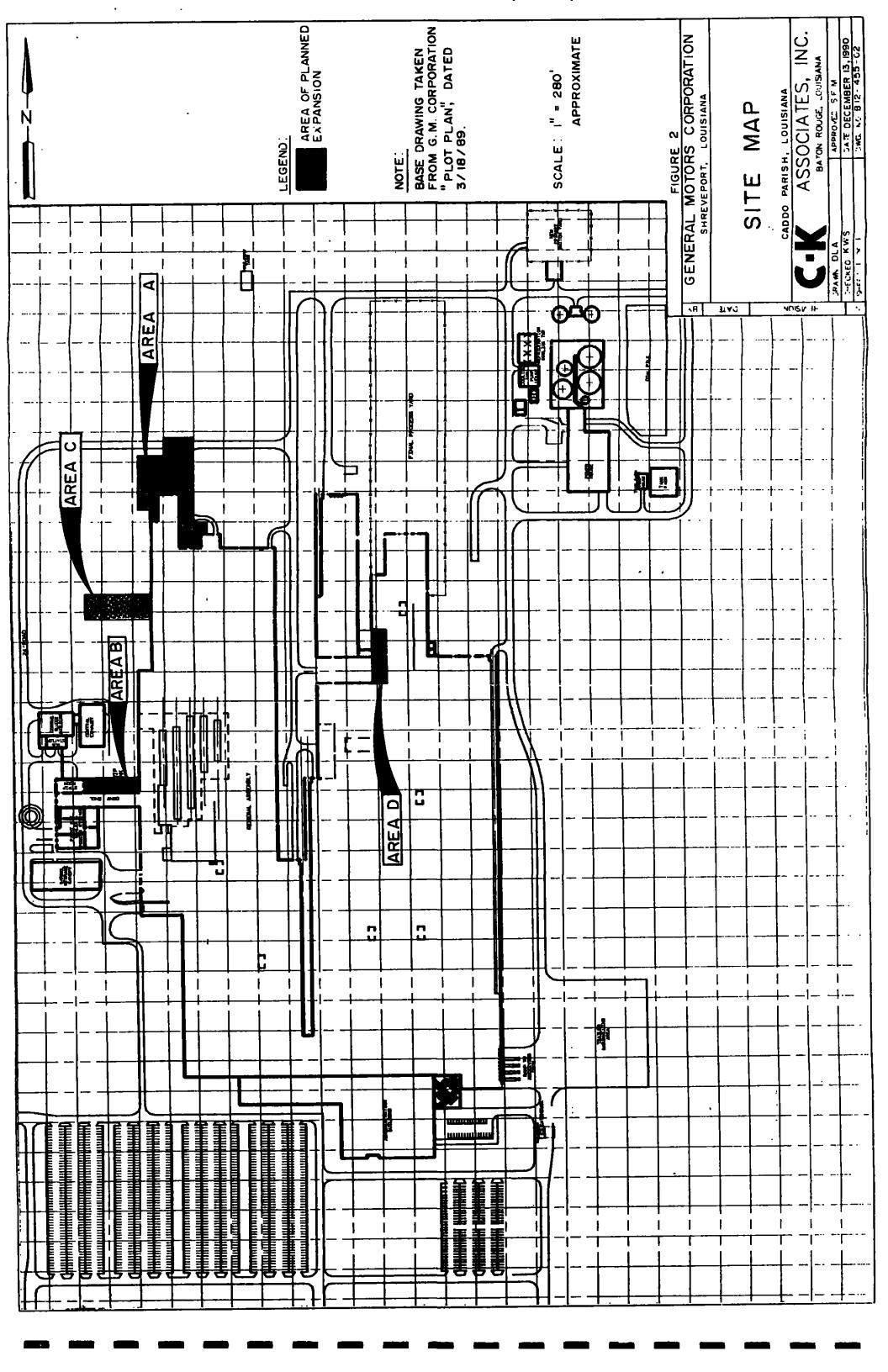
Phase II assessment activities conducted at Expansion Area D (Roll Test Booth) detected the presence of 2-Hexanone in soil samples collected at B-9 and B-10. The maximum depth in which 2-Hexanone was detected was ten feet below the ground surface.

From an analytical standpoint, the quantity of 2-hexanone found in sample no. 1001 is so close to the limit of detection, it should be discounted. Sample nos. 901 and 902 were found to contain a concentration of 2-hexanone at 4-5 times the detection limit used for this analysis. Due to the relatively low concentration of 2-hexanone in these samples and the limited occurrence, this compound may be a sampling or laboratory artifact.

Also, an Environmental Protection Agency document, "Laboratory Data Validation Functional Guidelines for Evaluating Organic Analyses," February, 1988, may be cited. It states that no positive sample result should be reported unless the concentration of the compound in the sample exceeds five times the amount in any blank.

Further interpretation of the 2-hexanone analysis is included as correspondence found in Appendix C.

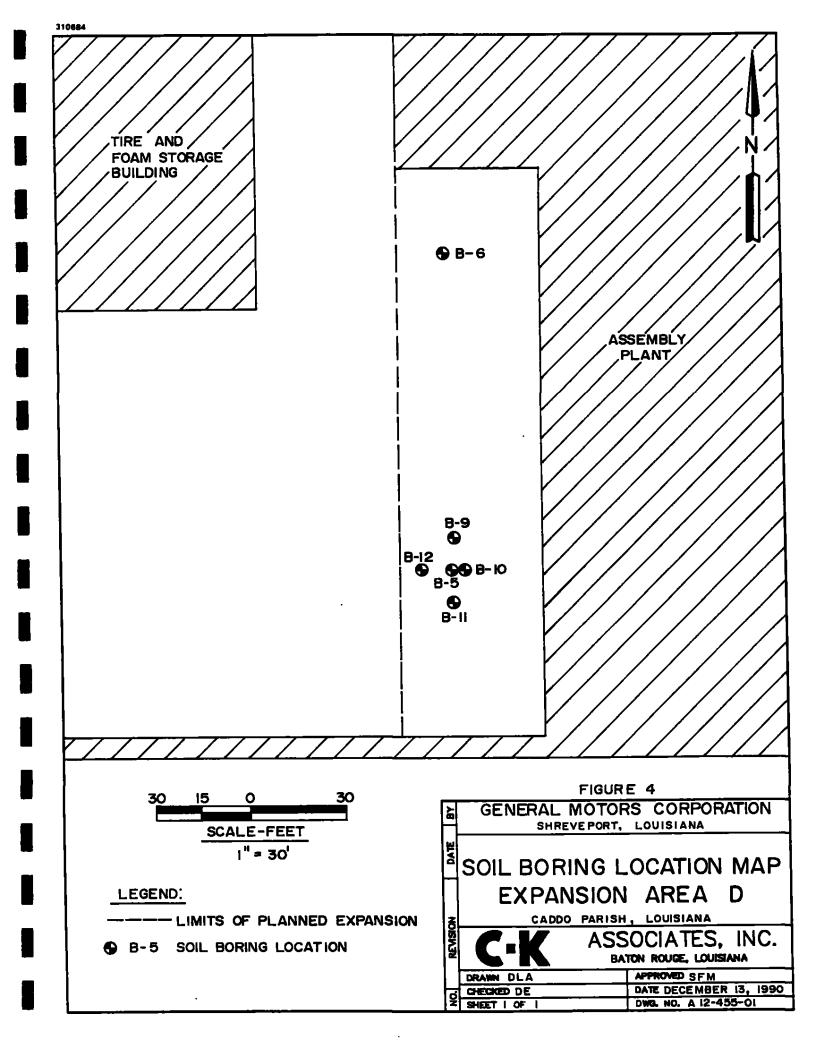


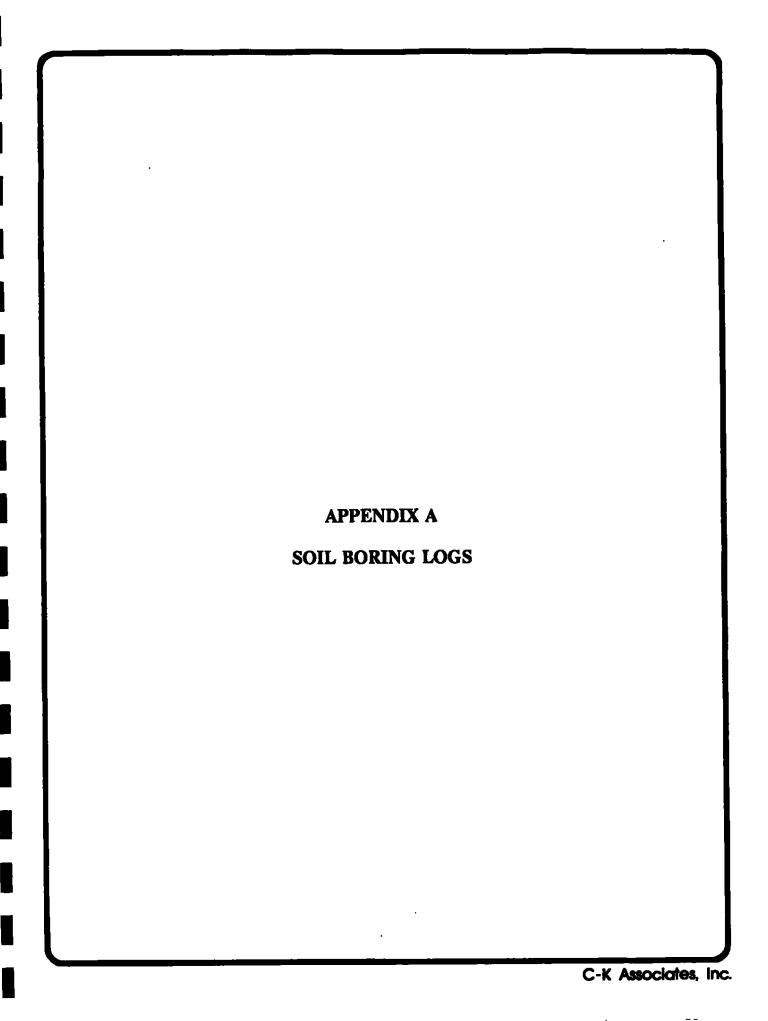


## **Reference Sheet**



**REF+16493** 





## SOIL BORING LOG

PROJE	CT NUM	BER	12-45	5-1	PROJECT NAME	General	Motors Co	orpo	ration			
BORIN	IG NUME	ER	B-9		COORDINATES				TE STARTE	Di i	12-10-	90
ELEVA	TION				GWL' DEPTH	DATE/TIME		-	TE COMPL			
C-K R	EPRESE	NTATIV	E ON SI	TE: Brad	Davis						_	
DRILL	ING ME	THODS	Но	llow stem a	auger			PA	GE 1		OF	2
OEPTH (FEET)	Recovery (90)	OVA (ppm)	Measured Consistency (TSF)		DESCRIPTION					REN	ARKS	
	30	10	4.0	Fill, re	d-brown, sandy	clay	-	nscs	Sample	No.	900	
	100	3	4.0	•			C	L				
- 5 <del>-</del>	90	0	4.0	Traces o	ght brown, sil f organic mate				Sample	No.	901 <sup>-</sup>	
•	90	0	>4.0	Iron oxi	de staining	,	C	L				
- 10	80	0	4.0	Hard, gr	ay-brown, silt	y clay,		$\dashv$	Sample	No.	902	<del></del>
	90	0	3.5	slickens				т.	-			
•	90	0	>4.0					_				
<b>-</b> 15 <b>-</b>	90	0	4.0	Hard, gr silt lam	ay-brown, clay	, fine san	d and		Sample	No.	903	
•	80	0	>4.0									
 - 20 -	90	0	1.0	Saturate	d zone at 19-2	0 feet	c	н				
	90	0	3.5						Sample	No.	904	
· -	90	0	4.0									
- 25-	90	0	>4.0	Dark gra	У		ļ		Sample	No.	905	
-	90	0	3.5									
<b>-</b>	90	0	4.0				1		Sample	No	906	

Laboratory samples were collected at 5-foot intervals

## SOIL BORING LOG

PROJE	CT NUM	AFR:	12-45	S_1	BBC ISCT NAME:	Concern No					<del></del>
						General Mc	tors Co				
			<u>0-7</u>	<del></del>	<del> </del>	DATE /TIME		$\overline{}$			
		NTATIV	E ON SI	TE: Brad	<del></del>	DATEFINE		UATE	COMPLETE	.O' 12.	-10-90
							<del></del>	PAGE	2	OF	2
DEPTH (FEET)	Recovery (90)	(w.zd.)	Measured Consistency (TSP)			ON	Zan A				
	90	0	4.0	Boring t	erminated_at_3	2_feet					
	BEPTH 130 30 10 11 11 11 11 11 11 11 11 11 11 11 11	BORING NUME ELEVATION: C-K REPRESE DRILLING ME (30)	ELEVATIONI C-K REPRESENTATIV DRILLING METHODS:  (60)  30  90  0	ELEVATION:  C-K REPRESENTATIVE ON SI  DRILLING METHODS:  HO  (ban)  90  0  4.0  4.0  Carsistency  (EEE.1)	BORING NUMBER: B-9 ELEVATION:  C-K REPRESENTATIVE ON SITE: Brad  DRILLING METHODS: Hollow stem  (BELLY AT 100	BORING NUMBER: B-9  ELEVATION:  C-K REPRESENTATIVE ON SITE: Brad Davis  DRILLING METHODS: Hollow stem auger  DESCRIPTION  30  90 0 4.0 Boring terminated at 3	BORING NUMBER: B-9 COORDINATES:  ELEVATION: GWL DEPTH DATE/TIME  C-K REPRESENTATIVE ON SITE: Brad Davis  DRILLING METHODS: Hollow stem auger  HA 30 90 0 4.0 Boring terminated at 32 feet  BORING NUMBER: B-9 COORDINATES:  DESCRIPTION  BORING NUMBER: B-9 COORDINATES:  DATE/TIME  DESCRIPTION	BORING NUMBER: B-9 COORDINATES: ELEVATION: GWL. DEPTH DATE/TIME C-K REPRESENTATIVE ON SITE: Brad Davis DRILLING METHODS: Hollow stem auger  TANGED OF THE COORDINATES:  DESCRIPTION  BORING TORRIDAD OF THE COORDINATES OF	BORING NUMBER: B-9 COORDINATES: OATE ELEVATION: GWL DEPTH DATE/TIME DATE C-K REPRESENTATIVE ON SITE: Brad Davis  DRILLING METHODS: Hollow stem auger PAGE	BORING NUMBER: 3-9 COORDINATES: DATE STARTED: ELEVATION: GWL-DEPTH DATE/TIME DATE COMPLETE C-K REPRESENTATIVE ON SITE: Brad Davis DRILLING METHODS: Hollow stem auger PAGE 2  ### A PAGE 2  ### A PAGE 2  ### A PAGE 3  ### A PAGE	BORING NUMBER: B-9 COORDINATES: OATE STARTED: 12-10- ELEVATION: GWL: DEPTH DATE/TIME DATE COMPLETED: 12- C-K REPRESENTATIVE ON SITE: Brad Davis  BRILLING METHODS: Hollow stem suger PAGE 2 OF  BLAD OF STARTED: 12-10- TO DATE STARTED: 12-10- TO DAT

## SOIL BORING LOG

	CT NUM		12-45	5-1	PROJECT NAME	General N	dotors Co	rpora	tion	<del></del>			
BORIN	G NUM	BER	B-10		COORDINATES:			DATE STARTED 12-11-90					
ELEVA	TION	-			GWL: DEPTH	DATE/TIME			COMPLET				
			E ON SI		Davis								
DRILL	ING ME	THOOS	Ho	llow stem	auger			PAGE	1	OF	2		
DEPTH (FEET)	Recovery (90)	MAC (mag)	Measured Consistency (TSF)		DESCRIPTION				RI	EMARKS			
_	50	10	4.0	Fill, re	d-brown, sandy	clay	·	S	ample No	- 1000	-		
	60	0	>4.0			•	CI	-			-		
5 -	70	2	4.0	Traces o	ght brown, sile f organic mate			s	ample No	. 1001	_		
	10	2	3.5	Iron oxi	de staining		CI	.  -			-		
101	90	0	3.5	777		<del></del>		-			<del>-</del>		
	90	0	>4.0	slickens	ay-brown, silty ided	y clay,	9	-	ample No	. 1002	-		
	90	0	>4.0				CI	<u> </u>			-		
15-	80	0	>4.0	Hard, graand silt	ay-brown, clay laminations	, fine sand		s	ample No	. 1003			
<u> </u>	90	0	>4.0								-		
20 -	90	0	1.0	Saturate	d zone at 19-20	) feet	CE		ample No	. 1004	]		
<u>.</u> -	90	0	>4.0								3		
<u> </u>	90	0	4.0								]		
25-	70	0	>4.0	Dark gra	У		ļ	s	ample No	. 1005			
;	80	0	>4.0								3		
	80	0	>4.0					S	ample No	. 1006	1		
NOTES	1												

Laboratory samples were collected at 5-foot intervals

## SOIL BORING LOG

99015	CT NUM	050.	12-45											
	G NUME		B-10		COORDINATES:	PROJECT NAME: General Motors Co								
ELEVA			B-10		GWL DEPTH					DATE STARTED: 12-11-90 DATE COMPLETED: 12-11-90				
		NTATIV	E ON SI	TE: Brad	Davis	DATE/TIME		DATE CO	MPLETED	12-11-90				
		THOOS		llow stem		<del></del>		PAGE		0F 0				
<b>—</b>						···	<del></del>		2	OF 2				
S OEPTH (FEET)	Recovery (90)	(uzid) 1870	Measured Omsistency (TSP)		DESCRIPTIO	МС	Serving States		REMA	RKS				
]	90	0	>4.0											
	80	0	>4.0							•				
35				Bor	ing terminated	at 35_feet .		San	nple No.	1007				
										•				
										•				
								ı						
										•				
		<u> </u>												
										•				
-														
POTES	'							<u></u>						
<u> </u>			·											

## SOIL BORING LOG

PROJE	CT NUM	BER	12-45	-1 PROJECT NAME	General Motors Co	rporation					
BORIN	G NUME	ER	B-11	COORDINATES	COORDINATES: DATE ST						
ELEVA	TION			GWL: DEPTH	DATE/TIME	DATE COMP	DATE STARTED: 12-11-90 DATE COMPLETED: 12-11-90				
			E ON SI								
DRILL	ING ME	THOOS	Но	low stem auger		PAGE	1 OF	2			
OEPTH C (FEET)	Recovery (90)	OM (mzti)	Measured Consistency (TSP)	DESCRIPTION		SCS STIMBOL	REMARKS				
•	20	0	2.0	Fill, red-brown, sandy c	1	_	No. 1100				
-	90	1	>4.0		C	<sup>L</sup>					
5 -	90	1	>4.0	Hard, light brown, silty Traces of organic materia	al	[	No. 1101				
-	80	2	>4.0	Iron oxide staining	C			•			
10 -	90	0	>4.0	Hard, gray-brown, silty slickensided	clay,						
-	90	0	>4.0		c	L Sample	No. 1102				
•	90	0	>4.0	HawA gray brown along		<del></del>					
15-	90	0	4.0	Hard, gray-brown, clay, and silt laminations	Tine sand	Sample	No. 1103				
•	80	0	4.0		,						
20-	80	0	4.0	Saturated zone at 19-20	feet	u	No. 444				
]. <u> </u>	90	0	4.0			sample	No. 1104				
	80	0	4.0	<b>.</b>							
25-	80	0	>4.0	Dark gray		Sample	No. 1105				
	80	0	>4.0								
	90	0	4.0		ĺ		No. 1106				

Laboratory samples were collected at 5-foot intervals.

## SOIL BORING LOG

PQ	DJECT N	UMBER:	12-45	5-1	PROJECT NAME	General I	lahara G				_		
_	RING NU		B-1		PROJECT NAME: General Motors Co				DATE STARTED: 12-11-90				
ELE	VATION	1			GWL: DEPTH					DATE COMPLETED: 12-11-90			
	REPR	SENTATI	E ON SI	TE: Brad	Davis			10-	IE COMPLE	20. 12-11-90			
		METHOOS		PA	GE 2	OF 2	$\dashv$						
B DEPTH	(FEET) Recovery	(tibbu)	Measured Consistency (TSP)		DESCRIPTION	USCS STMBOL		EMARKS					
	90	0	4.0								┪		
	90	0	4.0								7		
35	·			Bor	ing terminated	at 35 feet		-	Sample N	o. <u>1</u> 107	-		
	-												
											1		
											4 4 4 4		
F											444		
											1		
NOT	ESI												

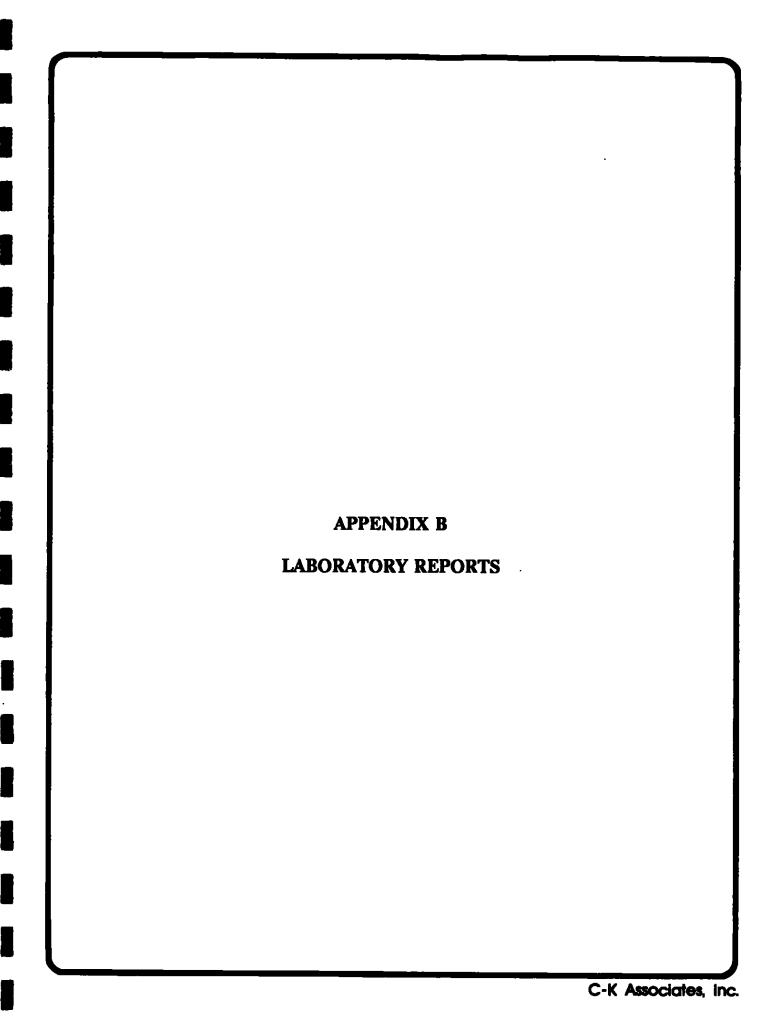
## SOIL BORING LOG

PROJE	CT NUM	BER	12-45	5-1	PROJECT NAME: General Motors Corporation						
BORIN	G NUME	ER	B-1	2	COORDINATES		0	ATE STARTED	12-11-	90	
ELEVA	TION				GWL: DEPTH	0	DATE COMPLETED: 12-11-90				
			E ON SI		Davis						
DRILLI	ING ME	THODS	Но	llow stem	auger		P	AGE 1	OF	2	
I ОЕРТН О (FEET)	Recovery (90)	(wid) VAO	Measured Constistency (TSF)		DESCRIPTIO	N .	USCS SYMBOL	REMA	ARKS		
	70	0	>4.0	Fill, re	ed-brown, sandy	clay	CL	Sample No.	1200	•	
	90	25	2.5							• •	
5 -	90	5	>4.0	Traces o	ght brown, sile of organic mate		Sample No.	1201	-		
- 	80	0	3.5	Iron oxi	Iron oxide staining	CL			- -		
 - 10	90	0	3.5	Hard, gr	ay-brown, silty	clay,	_	Sample No.	1202	<u>-</u>	
	90	0	>4.0	slickens	ided		CL			-	
-  -  -	90	0	4.0		<del></del>					_	
- 15-	90	0	4.0	Hard, gr and silt	ay-brown, clay,	, fine sand		Sample No.	1203		
	90	0	4.0							-	
20	90	0	>4.0	Saturate	d zone at 19-20	) feet		Sample No.	1204	-	
 - +	90	0	>4.0				СН			-	
┡ ┤ <b>-</b> ┽	90	<u> </u>	4.0	Dark gra	v					-	
25-	90	0	4.0	park dra	¥			Sample No.	1205	-	
-  -	90	0	>4.0			·				-	
100000	90	0	4.0	<u>-</u> -				Sample No.	206	<u>.</u>	
NOTES											

Laboratory samples were collected at 5-foot intervals.

## SOIL BORING LOG

PROJE	CT NUM	BER	12-45	5-1	PROJECT NAME	General Moto	rs Corp	Oration			
BORIN	B NUM	BER:	B- '	12	COORDINATES			DATE STARTED: 12-11-90			
ELEVA	TION				GWL: DEPTH	ATE COMPLETED: 12-11-90					
C-K R	EPRESE	NTATIV	E ON SI	TE: Brad	Davis	DATE/TIME		30			
DRILL	ING ME	THOOS	Но	llow stem			P	AGE 2 OF 2			
TH (T)	ery ()	12	red ency				SYMBOL				
DEPTH (FEET)	Recovery (90)	W (mdd)	Measured Consistency (TSF)		DESCRIPTIO		USCS STI	REMARKS			
• ·	90	0	4.0				·				
	90	0	4.0			٠					
35				Bo	ring_terminated	at_35 feet		Sample No. 1207			
<u> </u>											
, -								-			
• -								•			
1 1						•					
<b>,</b> †											
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SAMPLE ANALYSES

for

C K Associates 17170 Perkins Road Baton Rouge, LA 70810 ATTENTION: Ms. Laurie Pierce

December 14, 1990

WEST-PAINE

Aboratories Inc.

1919 GSPI AVE. - BATON ROUGE, LA 70820

C K Associates Baton Rouge, LA 70810 December 14, 1990 Inc. is documented for your designated
provided, is included in this report. Sample analysis was in accordance with Environmental Protection Agency protocol. receipt at West-Paine Laboratories, I Sample sample(s).

Test Methods for Evaluating Solid Waste, SW-846, July 1982 À.

<u>Parameter</u> 2-Hexanone

Method 8240

Documented results are shown on the following page(s).

Victor 3. Branchard, III General Manager

Baton Rouge, LA 70810 December 14, 1990 C K Associates

> 90/12/11 90/12/12 900 Sample Source: Date Collected: Date Received:

13:35 09:19 Time Collected: Time Received:

> Parameter (Units)

2-Hexanone

(mg/kg)

Quality Assurance Actual/Found Recovery Percent Results

N/A

N/A

<0.0>

90/12/12 00:00 JAS

Date/Time Analyst

87 101 80 1,2-Dichloroethane-d4: Toluene-dg: 4-Bromoflüorobenzene:

nal

WEST-PAINE Jaboratories INC. 7978 GSRI AVE. . BATON ROUGE, LA 70820

70810 Baton Rouge, LA December 14, 1990 C K Associates

Sample Source:

90/12/10 90/12/11 Date Collected:

Quality Assurance Actual/Found Recovery 10:50 09:01 **Percent** Time Collected: Time Received: Result Date Received: Parameter (Unit)

2-Hexanone (mg/kg)

N/A 0.24

N/A

90/12/11 00:00 JAS

Date/Time Analyst

86 91 75 1,2-Dichloroethane- $\mathbf{d_4}$ : Toluene-dg: 4-Bromoflüorobenzene:

7979 GSFLAVE. • BATON ROUGE, LA 70820 WEST-PAINE aboratories INC.

70810 Baton Rouge, LA December 14, 1990 C K Associates

> 902 Sample Source:

90/12/10 90/12/11 Date Collected: Date Received:

Time Collected: Time Received:

11:00 09:01

Date/Time Analyst

Parameter

(Unit)

Result

Recovery Percent

Quality Assurance Actual/Found 90/12/12 00:00 JAS

2-Hexanone (mg/kg)

0.21

102 122 94

1,2-Dichloroethane-d4:

Toluene-dg: 4-Bromoflüorobenzene:

N/A

N/A

70810 Baton Rouge, LA December 14, 1990 C K Associates

> 90/12/10 90/12/11 903 Date Collected: Sample Source: Date Received:

Time Collected: Time Received:

11:32 09:01

Quality Assurance Actual/Found Recovery Percent Result Parameter (Unit)

N/A < 0.05

2-Hexanone

(mg/kg)

90/12/11 00:00 JAS

N/A

Date/Time Analyst

> 95 100 79 1,2-Dichloroethane- $d_4$ : Toluene-dg: 4-Bromoflüorobenzene:

hal

19-3180

Baton Rouge, LA 70810 December 14, 1990 C K Associates

> 90/12/10 90/12/11 Date Collected: Date Received: Sample Source:

Time Collected: Time Received:

13:01 09:01

Parameter (Units)

2-Hexanone

(mg/kg)

Recovery Percent Results

Quality Assurance Actual/Found

Date/Time Analyst 90/12/11 00:00 JJB

N/A

N/A

Unable to analyze sample. could be recovered. NOTE:

Neither internal standards nor surrogate standards

19-3180



70810 baton Rouge, LA December 14, 1990 C K Associates

90/12/10 90/12/11 902 Sample Source: Date Collected:

13:25 09:01 Time Collected: Time Received: Date Received:

Quality Assurance Actual/Found Recovery Percent Result Parameter Unit

94 92 67 1,2-Dichloroethane-d4: Toluene-dg: 4-Bromoflüorobenzene:

N/A

< 0.05

2-Hexanone

(mg/kg)

N/A

90/12/12 00:00 JAS

Date/Time Analyst

hal

7979 GSRI AVE. . BATON ROUGE, LA 70620 WEST-PAINE aboratories INC.

70810 Baton Rouge, LA December 14, 1990 C K Associates

> 906 Sample Source: Date Collected:

90/12/10 90/12/11 Date Received:

14:15 09:01 Time Collected: Time Received:

Quality Assurance Actual/Found

Recovery

Result

Percent

Analyst

Date/Time

2-Hexanone (mg/kg)

Parameter

(Unit)

< 0.05

N/A

N/A

90/12/12 00:00 JAS

1,2-Dichloroethane-d4: Toluene-dg: 4-Bromoflüorobenzene:

88 85 59

hal

WEST-PAINE 7979 GSR! AVE. • BATON ROUGE, LA 70620 aboratories INC.

70810 Baton Rouge, LA 7 December 14, 1990 C K Associates

90/12/11 90/12/12 Sample Source: 1000 Date Collected: 90,

13:40

Date Received: Parameter

09:19 Time Collected: Time Received: Quality Assurance Actual/Found

Recovery Percent

Results

Date/Time Analyst

> 2-Hexanone (mg/kg)

(Units)

<0.05

N/A

N/A

90/12/12 00:00 JJB

1,2-Dichloroethane-d4: Toluene-d8:

4-Bromofluorobenzene:

98 100 75



70810 Baton Rouge, LA December 14, 1990 C K Associates

> 1001 Sample Source: Date Collected:

Time Collected: Time Received: 90/12/11 90/12/12 Date Received:

07:40 08:53

Date/Time Analyst	
Quality Assurance Actual/Found	
Percent Recovery	
Result	
Parameter . (Unit)	

90/12/11 00:00 JAS

N/A

N/A

0.09

2-Hexanone (mg/kg).

88 93 79 1,2-Dichloroethane-d4: Toluene-dg: 4-Bromoflüorobenzene: 19-3180



70810 December 14, 1990 Baton Rouge, LA C K Associates

> 1002 Sample Source:

90/12/11 90/12/12 Date Collected: Date Received:

07:50 08:53 Time Collected: Time Received:

Parameter (Unit)

Recovery Percent Result

> 2-Hexanone (mg/kg)

N/A < 0.05

N/A

90/12/11 00:00 JAS

Date/Time Analyst

Quality Assurance Actual/Found



70810 C K Associates Baton Rouge,

December 14, 1990

1003 Date Collected: Sample Source:

90/12/11 Date Received:

08:53 08:13 Time Collected: Time Received: Quality Assurance Actual/Found

Date/Time Analyst

2-Hexanone (mg/kg)

Parameter

(Unit)

Recovery Percent Result

90/12/11 00:00 JAS

N/A

N/A

< 0.05

1,2-Dichloroethane- $d_4$ :

90 90 89

Toluene-dg: 4-Bromoflüorobenzene:

7979 GSRI AVE. . BATON ROUGE, LA 70620 WEST-PAINE laboratories INC.

70810 Baton Rouge, LA December 14, 1990 C K Associates

> 1100 Date Collected: Sample Source:

90/12/11 90/12/12 Date Received:

08:53 09:40 Time Collected: Time Received:

Quality Assurance Recovery Percent Result

> 2-Hexanone (Unit)

(mg/kg)

Parameter

< 0.05

N/A

W/N

Actual/Found

90/12/12 00:00 JAS

Date/Time Analyst

> 1,2-Dichloroethane-d4: Toluene-d<sub>8</sub>: 4-Bromoflüorobenzene:

84 105 83

hal

1979 GSRI AVE . BATON ROUGE, LA 70820 aboratories INC. VEST-PAINE

70810 Baton Rouge, LA December 14, 1990 C K Associates

90/12/11 90/12/12 1101 Date Collected: Sample Source: Date Received:

Time Collected: Time Received:

09:55 08:53

Date/Time Analyst

> 2-Hexanone (mg/kg)

Parameter

(Unit)

Recovery Percent Result

Quality Assurance Actual/Found 90/12/11 00:00 JAS

1,2-Dichloroethane- $d_4$ :

83 91 67

Toluene-d<sub>8</sub>: 4-Bromoflüorobenzene:

< 0.05

A/N

N/A

hal



70810 Baton Rouge, LA December 14, 1990 C K Associates

90/12/11 90/12/12 1102 Date Collected: Date Received: Sample Source:

10:05 08:53 Time Collected: Time Received:

Date/Time Analyst Quality Assurance Actual/Found Recovery Percent Result Parameter (Unit) 90/12/11 00:00 JAS

N/A

N/A

< 0.05

2-Hexanone

(mg/kg)

82 91 67 1,2-Dichloroethane-d4: Toluene-dg:



70810 Baton Rouge, LA December 14, 1990 C K Associates

> 1108 Sample Source: Date Collected:

Date Received:

10:60 11:00 Time Collected: Time Received: 90/12/11

Percent

Quality Assurance Actual/Found N/A Recovery N/A Results <50 2-Hexanone Parameter (Units) (nd/Ir)

90/12/13 00:00 JAS

Date/Time Analyst

97 98 83 1,2-Dichloroethane- $d_4$ : Toluene-dg: 4-Bromoflüorobenzene: 19-3180

70810 Baton Rouge, LA December 14, 1990 C K Associates

90/12/11 90/12/12 Date Collected: Date Received:

1200

Sample Source:

12:12 09:19 Time Collected: Time Received: Quality Assurance Actual/Found Recovery Percent Result **Parameter** (Unit)

2-Hexanone

(mg/kg)

N/A < 0.05

90/12/12 00:00 JAS

N/A

Date/Time Analyst

> 88 106 80 1,2-Dichloroethane- $d_4$ : Toluene-d<sub>8</sub>: 4-Bromoflüorobenzene:

WEST-PAINE 7979 GSRI AVE. - BATON ROUGE, LA 70820

70810 Baton Rouge, LA December 14, 1990 C K Associates

1201 Sample Source: Date Collected:

90/12/11 90/12/12 Date Received:

12:20 09:19 Time Collected: Time Received:

Quality Assurance Actual/Found Recovery Percent

90/12/12 00:00 JAS

Date/Time Analyst

> 2-Hexanone (mg/kg)

Parameter

(Unit)

< 0.05

Result

N/A

K/Z

1,2-Dichloroethane-d4: Toluene-d<sub>8</sub>: 4-Bromoflüorobenzene:

84 105 79

70810 Baton Rouge, LA December 14, 1990 C K Associates

90/12/11 90/12/12 Date Collected: Date Received:

1202

Sample Source:

Quality Assurance 12:26 09:19 **Percent** Time Collected: Time Received: Parameter

Actual/Found N/A Recovery N/A Result < 0.05 2-Hexanone (mg/kg) (Unit)

90/12/12 00:00 JAS

Date/Time Analyst

> 75 94 69 1,2-Dichloroethane- $d_4$ : Toluene-dg: 4-Bromoflüorobenzene:

**EST-PAINE** 7979 GSRI AVE • BATON ROUGE, LA 70820 aboratories INC.

70810 Baton Rouge, LA 7 December 14, 1990 C K Associates

> 1400 Sample Source: Date Collected: Date Received:

90/12/11 90/12/12

13:50 09:19 Time Collected: Time Received: Quality Assurance Actual/Found Percent Recovery Results

2-Hexanone Parameter (Units) (ng/Ir)

. <50

N/A

N/A

90/12/13 00:00 JAS

Date/Time Analyst

87 92 78 1,2-Dichloroethane-d4: Toluene-dg: 4-Bromoflüorobenzene:



COMMENTS: VOSLO-016

### VOLATILE MATRIX SPIKE/MATRIX SPIKE DUPLICATE RECOVERY

(SOIL)

Sample No: 9	<u> 000719-0001</u>
--------------	---------------------

Level: (low/med): \_\_low

ĀĪ	IKE	SAMPLE CONCENTRATION	MS			
<del>-</del>		こういこせいできょうしょ			MS	QC
COMPOUND (u		CONCENTRALION	CONCENT	RATION	8	LIMITS
	g/kg)	(ug/kg)	(ug/	kg)	REC #	REC.
1.1-Dichloroethene 2	425	<100	20	21	83	<u> 59-172</u>
Trichloroethene 2	425	<100	22		91	62-137
Benzene 2	425	<100	22	78.	94	66-142
Toluene 2	425	<100	24		102	59-139
Chlorobenzene 2	425	<100	24	43	101	60-133
1A	'IKE	MSD CONCENTRATION	MSD %	8	QC	LIMITS
COMPOUND (1	g/kg)	(ug/kg)	REC #	RPD #	RPD	REC.
1.1-Dichloroethene					22	59-172
<u>Trichloroethene</u>					24	62-137
Benzene					21	66-142
Toluene			_		21	59-139
Chlorobenzene					21	60-133



# VOLATILE MATRIX SPIKE/MATRIX SPIKE DUPLICATE RECOVERY (WATER)

COMPOUND  1.1-Dichloroethene Trichloroethene Benzene Toluene	SPIKE ADDED (ug/L) 250 250	SAMPLE CONCENTRATION (ug/L) <10	MS CONCENT	RATION	MS % REC #	QC LIMITS
1.1-Dichloroethene Trichloroethene Benzene	(ug/L) 250	(ug/L)			8	•
1.1-Dichloroethene Trichloroethene Benzene	250				-	
Trichloroethene Benzene		<b>~</b> 10			<u> 1(L</u> O	REC.
Trichloroethene Benzene		<b>∠1</b> 0				
Benzene	250		268.		108	61-145
		<10	229.	69	92	71-120
Toluene	250	<10	233.	<u>78</u> _	94	75-130
	250	<10	254,	19	102	76-125
<u>Chlorobenzene</u>	250	<10	240.	77	96	76-127
	SPIKE	MSD	MSD			
	ADDED	CONCENTRATION	8	*	OC	LIMITS
COMPOUND	(ug/L)	(ug/L)	REC #	RPD #	RPD	REC.
1.1-Dichloroethene Trichloroethene	<del></del>				14 14	61-145 71-120
Benzene					13	75-130
Toluene					13	76-12
<u>Chlorobenzene</u>					11	76-127
# Column to be used  * Values outside of  RPD:out of  Spike Recovery:0	QC limits outside lim	its	es with a	n asteri	sk	
COMMENTS: <u>VOWLO-05</u>	9		<del></del>			

C - K ASSOCIATES, INC.

NC CHAIN OF CUSTODY AND ANALYTICAL REQUEST RECORD

	CLIENT: (reactal Mature Curb.	Eneral Ma	thes Can		P.O. NUMBER:	<u>.</u>	SAMPLED BY: Brad Pavis
	PROJECT NO.:	: 12	1-554-81		LABORATORY *:_	10: Lest-Paine	DATE: 12-10-90
	SAMPLE IDENTIFICATION	DATE	TIME	MATRIX	NO. OF CONTAINERS	PRESERVATIVE	ANALYSES AND INSTRUCTIONS
1	106	12-10-90	1050	Soi/	/	None	2- Hexanone
1,	405	12-10-80	1/00	Soil	,	Noas	2- Hexanone
٤,	903	12-10-90	//33	Sai!	,	Nane	2 Hexanone
て	404	12-10-90	12-10-90 1301		/	Near	2 - Hexanone
7	905	12-10-90 1325	1335	Sail	,	Nene	2 - Hexanone
٦,	906	12-10-90 H15	1415	Soil	/	None	2. Hexanone
- <del></del>						·	
							* All Samoles were Dlaced within
							7
							tro. of 44°
	Reimquished by: (Signalure)	(Signature)	77	ر ا	Dets /2-/0-90	007/	Paragrag by: (signalues
	Rolling and Mi Statutes	Stance	3		04.6 12.40-83	/700	Date

in our | Balan Rouge, | Ltate Charles, E Entevepert office. Ocal Causs \* Piesse send fesuits and invoice to the attention of

12-10-90

12106

Received for L'aboratory by: (Signature)

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Collection witnessed by: (Signeture, if Required)

Relinquished by: (Signeture)

Method of Shipment

SSOCIATES INC CHAIN OF CUSTODY AND ANALYTICAL REQUEST RECORD

C-K ASSOCIATES, INC.

Page \_\_\_\_\_\_

SAMPLED BY: Brad Vavis ANALYSES AND INSTRUCTIONS 13-11-80 HCXanenc 2 - Hexanone HLKANONE Hexanine HEXANIA > Hexanone 2- HEXAMONE 2- Hexanone 2- Hexanone 2 - Hexanone 2 Hexanone 2- Hexanone DATE: Becaived by: (Sloneture) , Υ Lust-Paine PRESERVATIVE None None None None Marc Non Mar VIAC Nonc None Noni Vonc LABORATORY \*: CLIENT: (T. Eners / Motors Corp. P.O. NUMBER: CONTAINERS NO. OF MATRIX 5011 5011 501/ 50!/ 5011 Soi! Sai / Soil 05/0/08-11-81 12-11-90 0955 0740 12-11-90 0813 080 08-11-81 12-11-90 0940 13-11-40 1030 12-11-90 0830 TIME 5001 03-11-61 12-11-90 CS#D 12-11-80 0850 8201 08-11-81 -55% -61 12-11-20 DATE PROJECT NO.:\_ IDENTIFICATION 1003 1005 1001 1004 7007 (00) 200/ (0/ 00// 101 4011

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A S S O C I A T E S I N C CHA

C-K ASSOCIATES, INC.

N CHAIN OF CUSTODY AND ANALYTICAL REQUEST RECORD

57											
Had Vavis	12-11-60	INSTRUCTIONS									
SAMPLED BY:	DATE:	ANALYSES AND INSTRUCTIONS	2 - Hexanor	2- HIXANIE	2. Huaning	2- HCKANONE					
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	LABORATORY *: LUCST - Park	PRESERVATIVE	Mone	None	None	Soding Thinkh					
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Metros	/25-/	TIME	1033	1040	1050 Soil	00//					
CACCEL	12-	DATE	12-11-80	9 <del>1,9/</del>	12-11-80	13-11-90 1100 Crafe					
CLIENT: GENERAL Metrol Card	PROJECT NO. :	SAMPLE IDENTIFICATION	5 0//	90//	70//	80//					
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C-K ASSOCIATES, INC.

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S - INC CHAIN OF CUSTODY AND ANALYTICAL REQUEST RECORD

P.O. NUMBER:

Motors Corp.

SAMPLED BY: BANGE

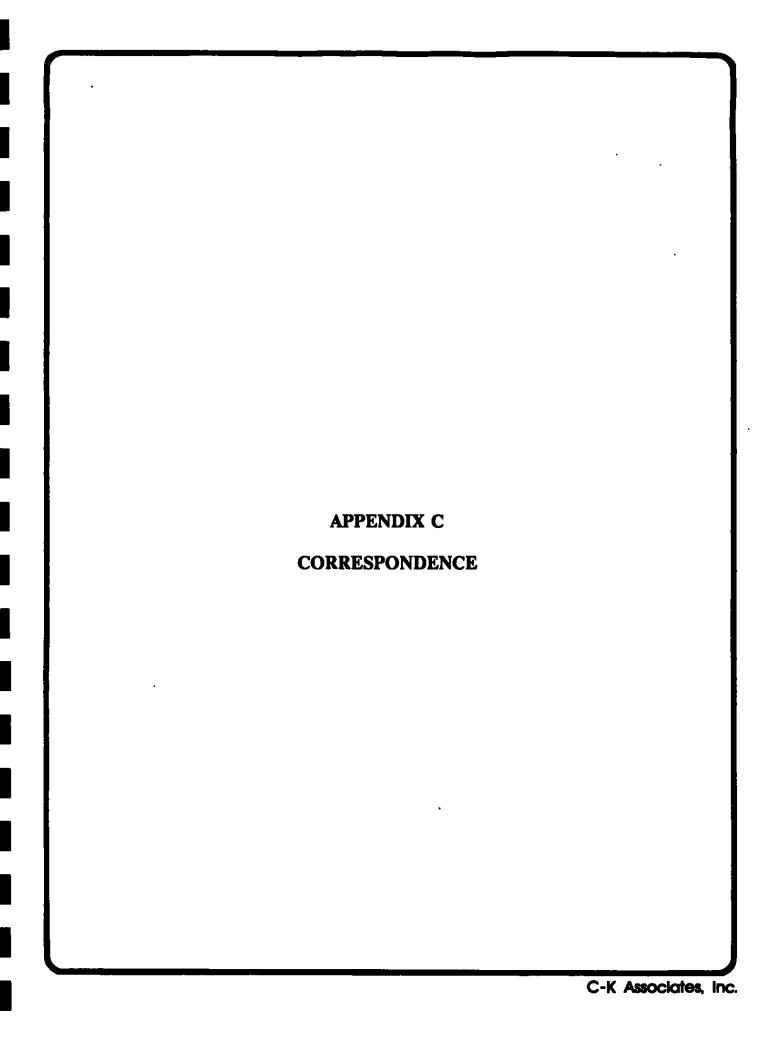
	PROJECT NO. :		17-55%-67		ABORATORN	LABORATORY .: Lucst - Paine	in DATE: 12-11-90
	SAMPLE IDENTIFICATION	DATE	TIME	MATRIX	NO. OF CONTAINERS	PRESERVATIVE	ANALYSES AND INSTRUCTIONS
4	1200	17-11-60	C/C/	50.1	\	Ninc	2- Hexanone
۴,		1		2011	,	Tool	2 - Hexamine
٩		3CC1 0b-11-81	3001	24.7	)	Your	2 - Hckaning
· 9<	•	12-11-90 1239	1239	56!	,	None	2- Heranone
1	1204	hhC1 06-11-81	hhC1	20.1	1	North	2 - 1-/cxangar
<u></u>	7305	12-11-90 1252	1.25.7	24:1	,	roy	2- Hexanine
<u>ئ</u> ې	1906	04-11-81	1300	ł	,	יאייינ	2- Flexanone
- 12		12-11-40	1308		,	Ning	2 HCKANINE
18		17-11-80	1335	Sai/	,	None	2- HKaning
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December 21, 1990

General Motors Corporation
Truck and Bus Manufacturing Division
Shreveport Plant
P. O. Box 30011
Shreveport, Louisiana 71130
Attn: Mr. H. Olin Desonier

Ref: West-Paine Laboratories, Inc.,

Soil Analysis of December 14, 1990; C-K Associates' Project No. 12-455-2

#### Dear Mr. Desonier:

The referenced soil samples were analyzed for 2-hexanone by Method 8240, "Test Methods for Evaluating Solid Waste," SW-846, July, 1982. This compound was detected in the following samples:

Boring No.	Sample No.	Sample Depth (ft)	2-Hexanone _(mg/kg)	Detection Limit (mg/kg)
B-9	901	5	0.24	0.05
B-9	902	10	0.21	0.05
B-10	1001	5	0.09	0.05

From an analytical standpoint, the quantity of 2-hexanone found in sample no. 1001 is so close to the limit of detection, it should be discounted. Sample nos. 901 and 902 were found to contain a concentration of 2-hexanone at 4-5 times the detection limit used for this analysis. Due to the relatively low concentration of 2-hexanone in these samples and the limited occurrence, this compound may be a sampling or laboratory artifact.

Also, an Environmental Protection Agency document, "Laboratory Data Validation Functional Guidelines for Evaluating Organic Analyses," February, 1988, may be cited. It states that no positive sample result should be reported unless the concentration of the compound in the sample exceeds five times the amount in any blank.

Mr. H. Olin Desonier December 21, 1990 Page 2

Please contact me if you have any questions.

Linda H. Grande

Very truly yours, C-K Associates, Inc.

Linda H. Grande, Ph.D. Environmental Specialist

LG/jn

cc: Mr. Bill Corbin



December 19, 1990

Certified Mail # P 094 288 267 Delivered by PAX # 504 342 6316

Mr. Leon Waller, Groundwater Protection Division, Louisiana Department of Environmental Quality P.O. Box 44272 Baton Rouge, LA 70804

RE: Interpretation of Analyses Results in 2nd Tier of 2hexanone analyses in GM Shreveport Assembly Facility Expansion Investigation

Dear Mr. Waller,

This letter provides information on additional borings performed at the GM Truck & Bus area "D" (attachment I). This area is part of the facility expansion slated to begin in September, 1991.

During Phase I of the Expansion Investigation, a composite boring located in area D showed a barely detectable level of 2-hexanone. Additional borings and analyses were performed as a second tier to determine if any contamination of the soil was present. The borings were spaced around the original composite boring (B-5) in order to "bracket" a possible pocket of contamination. The additional borings were analyzed at 2' and 5' intervals. Also, a new discrete boring (B-10) was made directly adjacent to the composite boring to verify the analyses results and more specifically quantify possible soil contamination.

All discrete borings showed below detection limits for all compounds tested except for insignificant levels of 2-hexanone (attachment II). GM believes these levels are insignificant for the reasons discussed in attachment III.

Mr. Leon Waller, December 19, 1990, page 2

Based upon the results of the completed Expansion Assessment, we believe that LDEQ-AQD should be allowed to proceed with the issuance of the air variances and our air permit.

If you have any questions, please call me (313) 456 6915.

Sincerely,

J B. Nachtman,

Sr Environmental Engineer, Truck & Bus Group, General

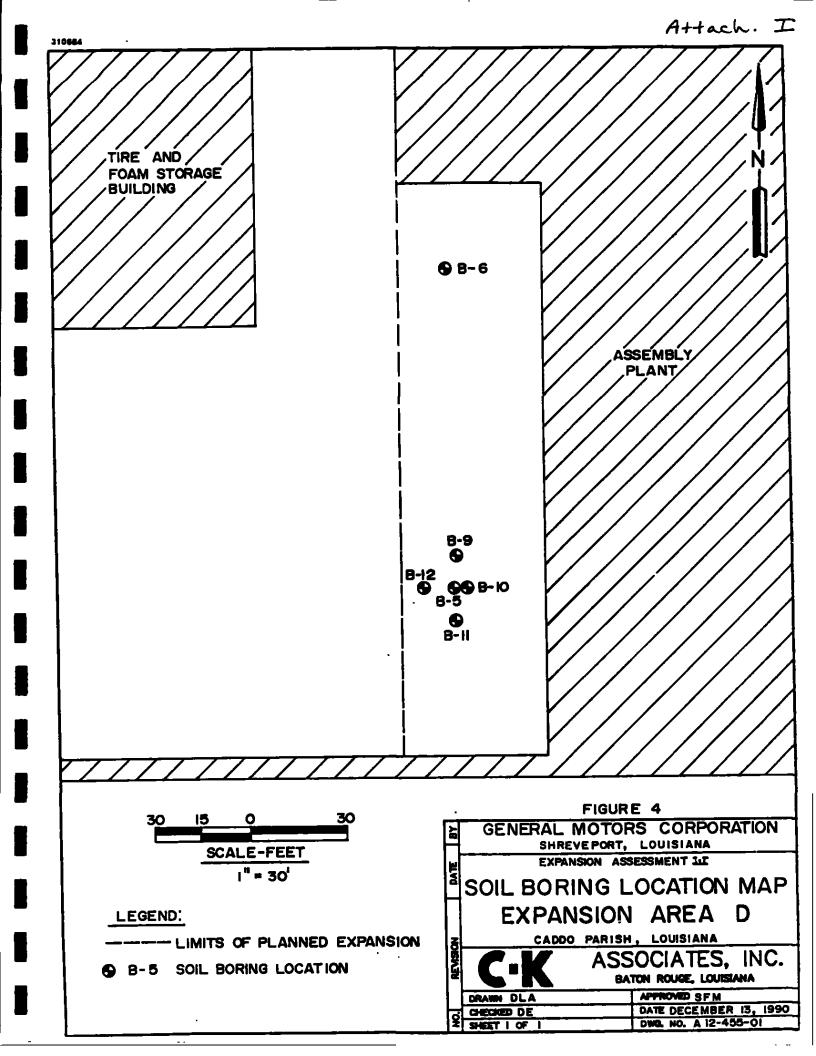
Motors Corporation

CC.

B. Davis, C-K Associates, Inc.

O. Desonier, GM T&B Shreveport

D. Dosher-Collard, LDEQ-AQD



### GENERAL MOTORS CORPORATION TABLE 1 SOIL ANALYSIS

Boring No.	Sample No.	Sample Depth (ft)	2-Hexanone (mg/kg)	Detection Limit mg/kg)
B-9	900	2	< 0.05	0.05
B-9	901	5	0.24	0.05
B-9	902	10	0.21	0.05
B-9	903	15	< 0.05	0.05
B-9	904	20	< 0.05	0.05
B-9	905	25	< 0.05	0.05
B-9	906	30	< 0.05	0.05
B-10 <sup>-</sup> \	1000	2	-	
B-10	1001	5	0.09	0.05
B-10 /	1002	10	< 0.05	0.05
B-10/	1003	15	< 0.05	0.05
B-11\	1100	2	< 0.05	0.05
B-11	1101	5	< 0.05	0.05
B-11 /	1102	10	< 0.05	0.05
B-12 \	1200	2	< 0.05	0.05
B-12	1201	5	< 0.05	0.05
B-12/	1202	10	< 0.05	0.05

	FIELD BLANKS	
Sample Number	2-Hexanone (ug/L)	Detection Limit (ug/L)
1108	<50	<50
1400	<50	<50

## RATIONAL FOR INTERPRETATION OF 2-HEXANONE ANALYSES

The rational for interpreting the level of 2-hexanone as insignificant is summarized below. Since each analyses point is a discrete sample, each concentration value can be assessed individually. As such, 3 out of 16 analyses showed values above the method detection limit (attachment II). One value, .09 mg/kg, we agreed can be disregarded outright as it is 4/100ths of a part per million (parts per million - ppm) above the detection limit. However, two analyses results showed .24 and .21 ppm in B-9;5' and B-9;10' respectively. How do we interpret these test results?

GM Truck & Bus believes that these two test results lie in the, "region of less-certain quantitation" as specified by the American Chemical Society and U.S. EPA. The basis for this conclusion is summarized as follows.

The U.S. EPA Office of Water addressed variability in low-level, near detection limit organic analyses results during the Thirteenth Annual Conference on Analyses of Pollutants of the Environment, held in Norfolk VA., on May 9-10, 1990 (please see attachment IV, Quantitation/Detection Limits for the Analyses of Environmental Samples). This presentation demonstrates EPA's view that analyses results that fall within 5 and 10 standard deviations from the Method Detection Limit (MDL) should be disregarded. The 10 standard deviation limit is also described as the Practical Quantitation Limit (PQL) and is considered a stringent standard to meet by most CERCLA (Superfund) contract laboratories.

The EPA uses PQLs which are recommended for volatile organic chemicals when it proposes MCLs for drinking water. Please note that our sampling matrix is soil, which is a more difficult matrix to quantify. The agency states that: "setting the PQLs in a range between 5 and 10 times the MDL achieved by the best laboratories is a fair expectation for most state and commercial laboratories" (50 FR 46907). The performance evaluations made by EPA showed that 80% of the labs could measure within ±40% of the true concentration. Therefore, even at the PQLs (10x the MDL), over 20% of the "good" labs would not be expected to obtain results within the ±40% of the specific component! At concentrations levels below the

PQL, performance of even the best of "good" contract laboratories deteriorated rapidly (see attachment IV).

Also, the US EPA in the Hazardous Site Evaluation Division published a guidance document on February 1, 1988, called the Laboratory Data Validation Functional Guidelines for Evaluating Organic Analyses (a copy of which was given to you in our meeting of December 17, 1990). This document, essentially an EPA SOP (Standard Operating Procedure) is used by EPA for general guidance in the technical interpretation of organic analytical test data. These guidelines recommend that if a sample result is greater than the Contract Required Quantitation Limit (CRQL), but is less than 5x the blank result (10x for some volatiles such as 2-butanone), the sample results would be qualified as non-detects (Please refer to page 13, regarding, "Blank Qualification Guidelines").

Applying these EPA guidelines to the two data points in question, the 5x rule would apply to the compound 2-hexanone.

#### 5x Rule

Blank Result	50	ug/L
CRQL	10	ug/Kg
Sample Result	240	ug/Kg
Qualified Sample Result	240	U ug/Kg

Using the 5x rule, sample results less than 250 (or 5 x 50) would be qualified as non-detects. The designation "U" means that the material was analyzed for, but was not detected. The associated numerical value is the sample quantitation limit. The CRQL of 10 ug/Kg was taken from the <u>US Contract Lab Program Scope of Work for Multi-Media/Multi-Concentrations</u>, revised 8/87 (attachment V).

Please note the blanks used in this qualification are water blanks. The sample matrix is soil. This adds an additional safety factor in the analyses, as the blank result is biased downward, although the amount of bias cannot be quantified with this data.

Attack IV

## QUANTITATION/DETECTION LIMITS FOR THE ANALYSIS OF ENVIRONMENTAL SAMPLES

#### I. INTRODUCTION

Analytical technology continues its unrelenting pace to develop methodology to lower the concentration limits at which the analytes can be measured. Picogram ( $10^{-12}$  grams) quantities are commonly reported as new detector systems for gas and liquid chromatography are developed. Advances in mass spectrometry are leading to lower levels of quantitation. For example, ion trap mass spectrometers and inductively coupled plasma-mass spectrometry (ICP-MS) are some highly sensitive techniques, which are becoming more commonly used for organic and elemental determinations respectively and capable of detecting subnanogram ( $<10^{-9}$  gram) quantities. The statement following depicts the situation that we are encountering:

"... the number of compounds detected in a sample of water is related to the detection level. As the detection level decreases an order of magnitude, the number of compounds detected increased an order of magnitude. Based on the number of compounds detected by current methods, one would expect to find every known compound at a concentration of 10-12 g/L or higher." - Dr. William T. Donaldson (EPA Athens Laboratory)

As the regulated community is required to perform within the level of increasingly restrictive compliance limits, the analytical chemist must emphasize to the public that all measurement data have an associated uncertainty interval(1). This information becomes critical as measurements are made approaching the lowest analytical capability of a given procedure.

### IV. PRACTICAL QUANTITATION LIMITS (POL) AS A MEANS OF IDENTIFYING MEASUREABLE CONCENTRATIONS

Many observations for organic toxic pollutants are below the MDLs, thus creating difficulties in developing effluent limitations guidelines and permit limits. In such instances where analytical and effluent variability cannot be determined, only those concentrations above quantifiable levels (17) should be considered. It should also be recognized that there is a fundamental difference between detection and quantitation limits. Unfortunately these terms are too often misused as being synonymous. EPA has developed a method for establishing such quantifiable numerical limits for its proposed drinking water standards (50 FR 46902) and for its proposed organic toxicity characteristic (51 FR 21652), designated as the practical quantitation limit (PQL). EPA has developed this concept of a PQL for specific analytical methods and lists of chemicals.

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### A. RECOMMENDED PRACTICAL QUANTITATION LIMITS COMPARED TO METHOD DETECTION. LIMITS

The EPA used PQLs which are recommended as 10 times the MDL for selected volatile organic chemicals when it proposed MCLs for drinking water. The Agency states that: "setting the PQLs in a range between 5 and 10 times the MDL achieved by the best laboratories is a fair expectation for most state and commercial laboratories" (50 FR 46907). At the PQLs chosen by EPA for this rulemaking, its performance evaluation studies indicate that 80% of the EPA and State laboratories in its water program evaluation studies could measure within ±40% of the true concentration. This was the basis for setting the PQL at 10 times the MDL. This is not a very high standard of performance as admitted by the Agency in the preamble to this proposed regulation. Thus, even at the PQLs, over 20% of the "good" laboratories would not be expected to obtain results within ±40% of the concentration of a specific component. At concentration levels below PQL, performance of even the best of "good" laboratories deterioriates rapidly.

## B. PRACTICAL QUANTITATION LIMITS IN REAL MATRIX SAMPLES REFLECT EFFECT OF MATRIX INTERFERENCE

A recent presentation(12) described a study evaluating Method 8020, which is a gas/liquid chromatography procedure in SW-846 "Test Methods for Evaluating Solid Wastes, Physical Chemical Methods" for the determination of low concentrations of toluene, benzene, and xylenes in real matrix groundwater samples. The round robin study involved 20 commercial laboratories. Method 8020 lists the practical quantification limits for all three compounds as 2.0  $\mu$ g/L. The PQLs derived from results achieved by the laboratories in this study are much higher. The PQLs at which 80% of the laboratories could achieve a recovery within  $\pm$ 40% of a true value from this study are 7.5  $\mu$ g/L for benzene, >20  $\mu$ g/L for toluene, and 18.5  $\mu$ g/L for total xylenes. It is clear that the Method 8020 published PQLs are seriously underestimated when applied to this groundwater matrix and for these 20 laboratories.

### V. PROPER TREATMENT OF THE DATA CAN AVOID MISREPRESENTATION OF THE FACTS

#### A. RULES FOR THE USE OF SIGNIFICANT NUMBERS

Despite the wide attention given to numbers for quantitative and qualitative limits the improper use of rules for use of significant numbers goes virtually unnoticed. As measurements are required more and more frequently to be made at decreasing concentrations, the relative analytical variability and uncertainty can increase substantially and the need to understand and recognize significant data is essential. Horwitz et al (22) reviewed data from over 50 independent Association of Official Analytical Chemists (AOAC) interlaboratory collaborative programs covering numerous AOAC drug and pesticide studies. The analytical methods covered were chromatography, atomic absorption spectrometry, absorption spectrometry, polarography, and biossay. In Figure 1 the % variation is expressed as powers of 2 with the mean concentration expressed as powers of 10. A convenient reference point is that at 1 ppm the variation is 16%. variation was found to double for each decrease of concentration by 2 orders of It is important to note that this curve is independent of the analyte or analytical technique that was used to make the measurements. These relationships should also apply to environmental levels of measurement as well.

Analytical chemists must always emphasize to the users of the data that the single most important characteristic of any result obtained from one or more analytical measurements is an adequate statement of its uncertainty interval. Often in legal judgments there is an attempt to dispense with uncertainty and try to obtain unequivocal statements; therefore, an uncertainty interval must be clearly defined in cases involving litigation and/or enforcement proceedings. Otherwise, a value of 1.001 without a specified uncertainty, for example, may be viewed as legally exceeding a permissible level of 1(7).

The analytical inclusion of only significant numbers is vital to the accurate interpretation of data. Scientific personnel are not exempted from the tendency to retain all values, no matter how divergent or suspect they may be. One of the principles of handling the data of physical and chemical measurements is that a numerical result by itself should give an approximate idea of the precision of the value as indicated by the number of significant figures used in expressing the value. An inaccurate representation of significant figures may give one an impression nearly as erroneous as from an inaccurate value. Misuse of significant figures can cause reporting violations when indeed the measured value does not exceed the limit. Adherence to proper expression of significant numbers is especially important when permit limits are near the limit of quantitation for the procedure and its relative uncertainties are large.

The number of significant figures reported as a result of a scientific measurement depends on establishing previously the relative precision with which the measurement can be made as shown in Table II(11). In considering the proper use of significant figures for regulatory reporting, it is imperative that significant figures start at the laboratory bench and be adhered to by anyone who further treats or handles the data. Otherwise, false conclusions and misunderstanding will develop and possibly lead to serious consequences.

Several observations can be made regarding the probability distribution shown in Figure 3.

Observation #1: Only a small percentage of the total analyses may give the best estimate of the true value.

Observation #2: One-half the measurements are above the mean and one-half of the measurements are below the mean. Therefore, if the mean is some effluent trigger concentration above which a plant would be violating its permit, the plant would be failing one-half the time, if these data were treated as having no uncertainty.

Observation #3: The measured concentrations shown in Figure 3, 99.7% of the reported values would fall between plus or minus  $3\sigma$  of the mean concentration; therefore, it can be seen that the  $\sigma$  of a determination is a very fundamental property of a distribution which must be used in evaluating data which contains uncertainty.

#### B. THE APPLICATION TO REGULATORY LIMITS

In order to translate this general probability distribution to real-world examples, Figures 4 through 7 were generated assuming different analytical uncertainty in the random errors. All figures were generated for the measurement of an effluent sample containing 100  $\mu$ g/L of the target analyte. Figure 4 shows the distribution of measured concentrations when the analytical uncertainty produces a value of  $1\mu$ g/L for  $\sigma$ ; Figure 5 shows the distribution of measured concentrations when the analytical uncertainty produces a value of 10  $\mu$ g/L for  $\sigma$ : Figure 6 shows the distribution of measured concentrations when the analytical uncertainty produces a value of 30  $\mu$ g/L for  $\sigma$ ; and Figure 7 shows the distribution of measured concentrations when the analytical uncertainty produces a value of 100  $\mu$ g/L for  $\sigma$ . The probability distribution for the last case has been truncated at 0  $\mu$ g/L since negative values of concentration are meaningless.

These four cases show clearly the impact of determinations which are carried out with different amounts of analytical uncertainty. Unfortunately, regulations are written as if data were being obtained with an uncertainty less than that shown in Figure 4. Permits which give a specific limit for a certain compound, fall into this category. However, the analytical data which are being obtained by a typical environmental laboratory for the analysis of reagent water are most likely analytical data obtained with the uncertainty shown in Figures 6 or 7. Figure 6 describes most analytical data obtained using EPA Methods 624 and 625 when the measured concentration is ten times higher than the method detection limit determined in reagent water. Figure 7 describes most analytical data obtained using EPA Methods 624 and 625 when the measured concentration is equal to the method detection limit which can be the case if the sample or sample extract must be diluted due to interfering substances. The concern is that the probability distribution summarized in Figure 6 is used by the Environmental Protection Agency to characterize data obtained by analytical laboratories for effluent analyses. However, these data represent a best case, since method detection limits for Methods 624 and 625 are derived from the analysis of reagent water. Reagent water data should not necessarily be used to determine the random error associated with all plant effluents which may contain relatively high levels of inorganic salts, and unregulated organic compounds which may interfere with these methods.

#### VII. RECOMMENDATIONS

There is a LOD or MDL which can be determined for every analyte in every matrix below which it is not possible to reliably ascertain that an analyte is present or absent. There is also a concentration range above the LOD or MDL where it is possible to qualitatively establish the presence of an analyte, but the concentration cannot be accurately and reliably quantified. It is also not practical to determine precisely the LOD or MDL for all analytes, in every matrix, and at all laboratories. All regulatory programs must recognize these As a practical solution to this problem, every method should have published practical quantification limits (PQLs) which are at least media (water/soil) specific. Many of these PQLs have been published by media, and for most analytes these PQLs are representative of levels that can be achieved at However, there should also be procedures for most commercial laboratories. determining matrix specific detection and quantitation limits. Unfortunately it is not possible to analyze a large enough universe of matrices to establish generalized quantitation limits for comparison with regulatory levels. approach must be established which will preserve the utility of published PQLs as guidance, while recognizing the significant number of compliance limits which are below their respective PQLs and thus require a variance procedure.

If a laboratory determines that it can not meet published detection and quantitation limits in their sample matrix, they should be allowed to measure these levels using established procedures which include mandated QA/QC requirements. These levels would then be used as reporting limits. If the quantitation limit, so established, is above the regulatory level, the compound would be considered to be in compliance until such a time that a level above the quantitation limit is measured. This assumption of compliance would apply whether or not the quantitation limit were a published PQL or a measured quantitation limit. EPA would also determine the frequency that these published PQLs would be re-evaluated pending method and equipment improvement. In some cases the Agency has suggested that a facility may petition for such a variance (24).

We also recommend that the EPA establish uniformity among the various regulatory programs for the determination of the method detection limit. Although the definition is essentially the same, the number of replicates and blanks may be different, therefore, the calculation is effected. This can further compound the current state of confusion in understanding and applying quantitation and detection limits. The corresponding quantitation limit should be established at five to ten times the MDL or substantially higher as the matrix would dictate (19). The use of such factors, however, must be used with extreme care as the method variability may well be underestimated by most laboratories (17). EPA recognized this need for consistency in its Report to Congress in CWA Section 518. It was reported that analytical methods are sometimes unnecessarily different for similar sample matrices, target analytes and data quality The Agency should move to greater method uniformity and more consistency in the use of quantitation and detection limits and use the concepts in regulatory compliance situations.

18. USEPA "Statistical Analysis of Ground Water Monitoring Data at RCRA Facilities, Interim Final Guidance" Office of Solid Waste Management Division, February, 1989, Section 8.

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- 19. "Test Methods for Evaluating Solid Wastes, Physical/Chemical Method" Third Edition, 8010-10, USEPA Office of Solid Waste, Revision I, December, 1987.
- 20. Parr. J., K. Carlberg, and G. Ward, "Reporting of Low Level Data for U.S. Environmental Protection Agency Needs", Presented at: Third Chemical Congress of North America Symposium in Honor of W. E. Harris, June 8, 1988.
- 21. Method Detection Limits and Practical Quantitation Limits for Incinerator Ash Matrices-Interlaboratory Study. Prepared for the Office of Solid Waste, USEPA, Washington, D.C. Prepared by the Analytical Chemistry Committee, Hazardous Waste Treatment Council, December 22, 1989.
- 22. Horwitz, W., Anal. Chem., 1982, 54 (1), 67A 76A
- 23. "Handbook for Analytical Quality Control in Water and Wastewater Laboratories" EPA-600/4-79-019, Chapter 7, Environmental Monitoring and Support Laboratory, USEPA Office of Research and Development, Cincinnati, Ohio.
- 24. 54 Federal Register 26603, June 23, 1989.

# DR. WILLIAM T. DONALDSON (EPA ATHENS LABORATORY)

"... THE NUMBER OF COMPOUNDS DETECTED IN A SAMPLE OF WATER IS RELATED TO THE DETECTION LEVEL. AS THE DETECTION LEVEL DECREASES AN ORDER OF MAGNITUDE, THE NUMBER OF COMPOUNDS DETECTED INCREASED AN ORDER OF MAGNITUDE. BASED ON THE NUMBER OF COMPOUNDS DETECTED BY CURRENT METHODS, ONE WOULD EXPECT TO FIND EVERY KNOWN COMPOUND AT A CONCENTRATION OF  $10^{-12}$  G/L or higher."

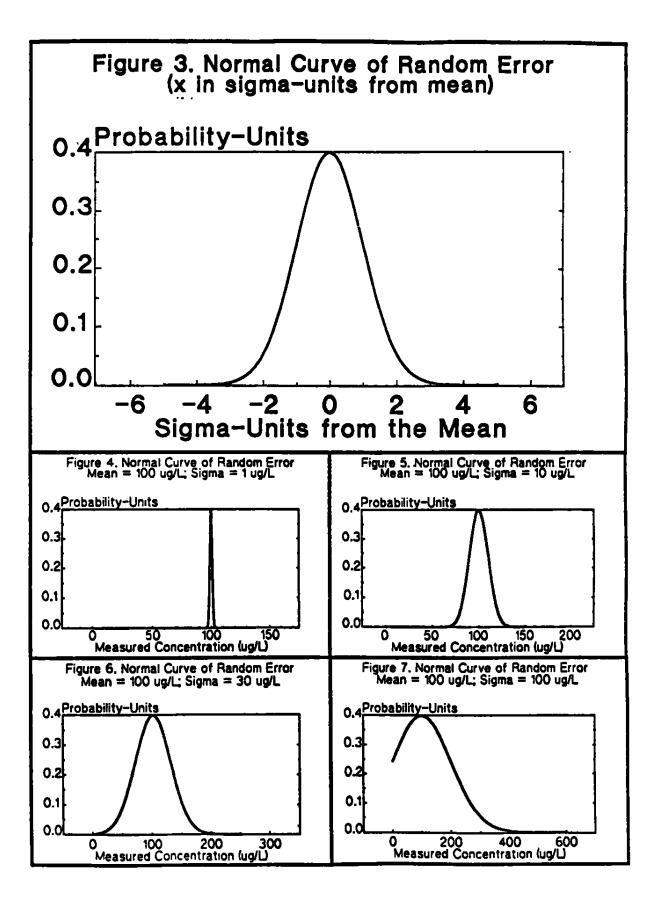
(1) Laboratories receive performance evaluation samples in which a limited number of concentrations are analyzed and the samples do not have matrix interferences as might actual samples;

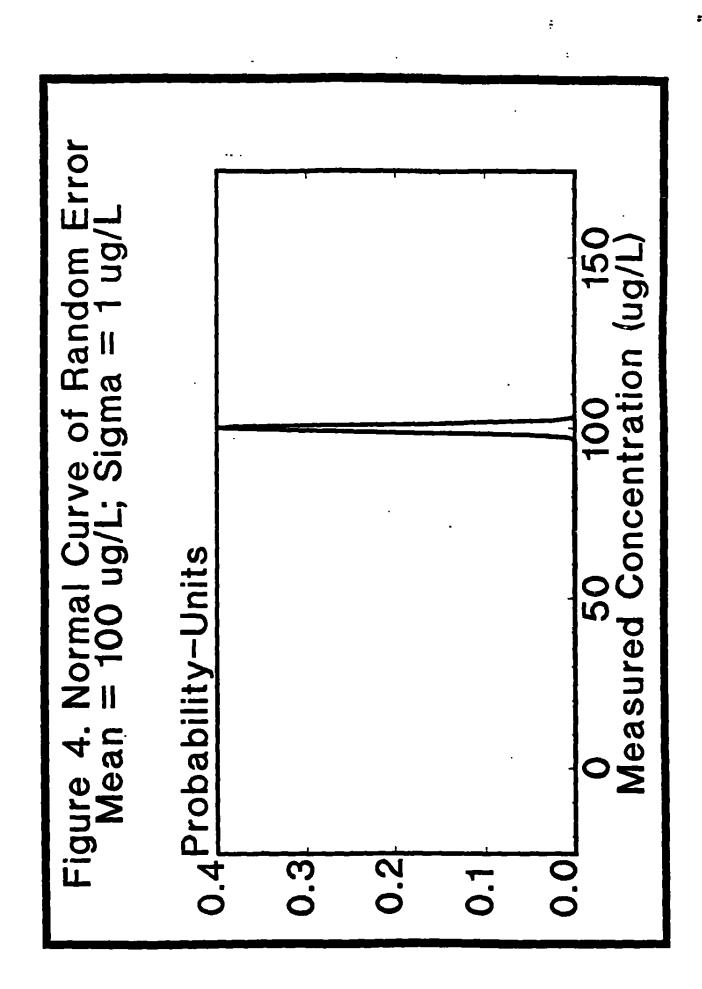
- (2) PQLs are based on EPA and State laboratory data which are considered to be representative of the best laboratories, but not all laboratories; and
- (3) Samples are analyzed under controlled ideal testing conditions which may not be representative of routine practices.

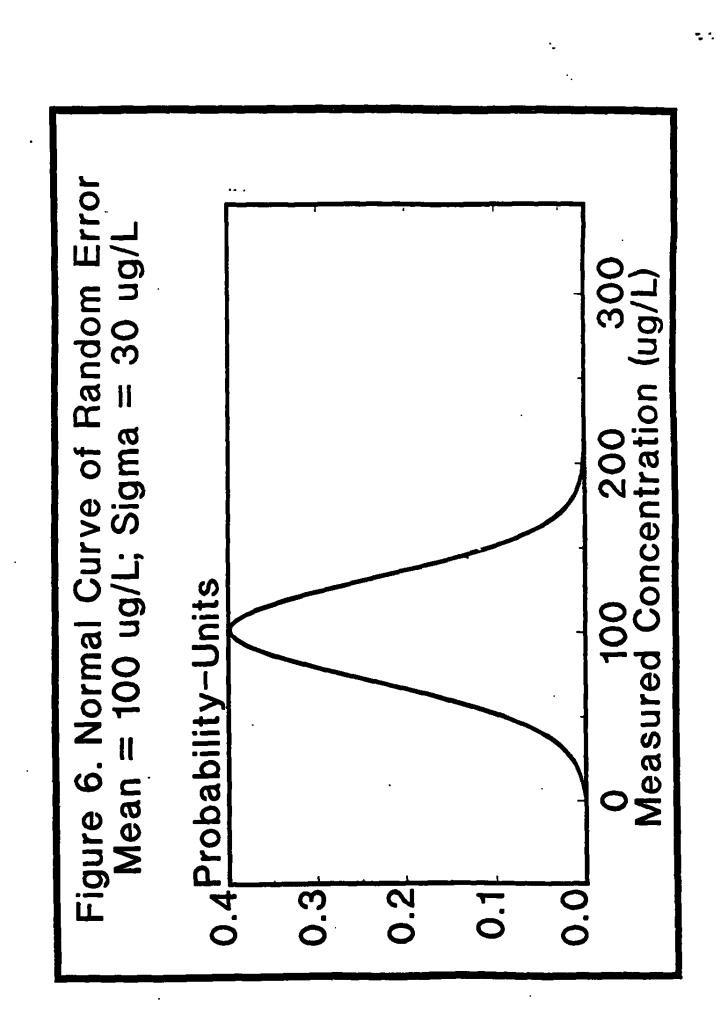
For these reasons, the PQL represents a relative stringent target for routine performance. (52 <u>Federal Register</u> 25699).

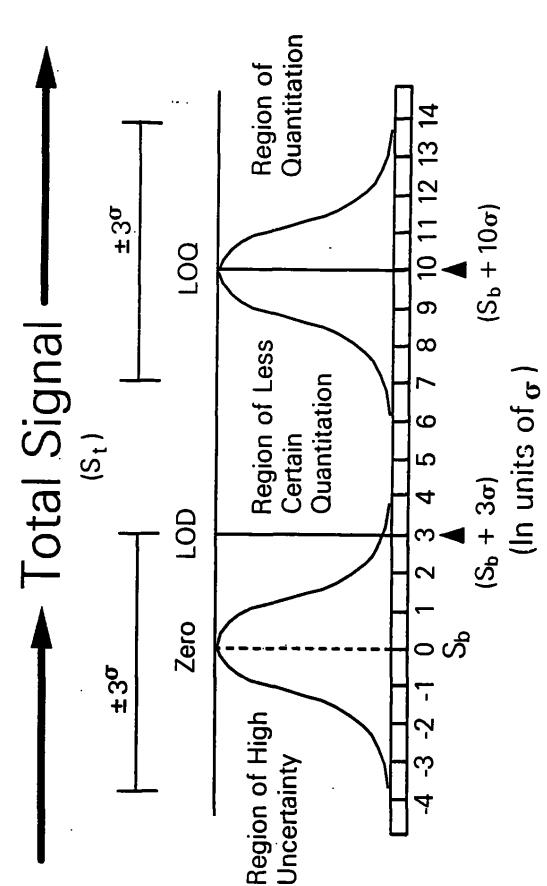
# Comparison of Reportable Significant Figures As A Function of Relative Precision

	Significant <u>Figures</u>	Example	
Precision (%)		Calculated	Reported
±0.001 to ±0.01	5	54.8149	54.815
$\pm 0.01$ to $\pm 0.1$	4	54.8149	54.81
±0.1 to ±1	3	54.8149	54.8
± 1 to ±10	2	54.8149	55
±10 to ±30	$\overline{1}$	54.8149	5 x 10 <sup>1</sup>









REPRINTED WITH PERMISSION FROM ANAL CHEM. 1983, 55, 22 10-22 18. COPYRIGHT 1983, AMERICAN CHEMICAL SOCIETY. FIGURE 2, RELATIONSHIP OF LOD AND LOQ TO SIGNAL STRENGTH

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# Target Compound List (TCL) and Contract Required Quantitation Limits (CRQL)\*

			Quantitation Limits**	
			Water	Low Soil/Sedimenta
	Volatiles	CAS Number	ug/L	ug/Kg
1.	Chloromethane	74-87-3	10	10
2.	Bromomethane	74-83-9	10	10
3.	Vinyl Chloride	75-01-4	10	10
4.	Chloroethane	75-00-3	10	10
5.	Methylene Chloride	75-09-2	5	5
6.	Acetone	67-64-1	10	10
7.	Carbon Disulfide	75-15-U	5	5
	1,1-Dichloroethene	75-35-4	5	5
9.	l, l-Dichloroethane	75-34-3	5	5
10.	1,2-Dichloroethene (total	) 540-59-0	š	5
11.	Chloroform	67-66-3	5	5
	1,2-Dichloroethane	107-06-2	5	, 5
	2-Butanone	78-93-3	10	10
	1,1,1-Trichloroethane	71-55-6	5	5
15.	Carbon Tetrachloride	56-23-5	5	5
16.	Vinyl Acetate	108-05-4	10	10
17.	Bromodichloromethane	75-27-4	5	5
18.	1,2-Dichloropropane	78-87-5	5	5
19.	cis-1,3-Dichloropropene	10061-01-5	5	Š
20.	Trichloroethene	79-01-6	5	5
21.	Dibromochloromethane	124-48-1	5	5
22.	l,i,2-Trichloroethane	79-00-5	5	5
23.	Benzene	71-43-2	5	Š
24.	trans-1,3-	<del>-</del>	5	. 5
	Dichloropropene	10061-02-6	•	•
25.	Bronoform	75-25-2	5	5
- 34	A-Mathyl-2-pentanone	108-10-1	10	10
27.	2-Hexanone	591-78-6	10	10
28.	recracuroroethene	127-18-4	5	5
29.	<del>-</del> -			
	Toluene 1,1,2,2-Tetrachloroethane	108-88-3	5	5

### GENERAL MOTORS CORPORATION TABLE 1 ANALYTICAL DATA

Boring No.	Sample No.	Sample Depth (ft)	2-Hexanone (mg/kg)	Detection Limit (mg/kg)
B-9	900	2	< 0.05	0.05
B-9	901	5	0.24	0.05
B-9	902	10	0.21	0.05
B-9	903	15	< 0.05	0.05
B-9	904	20		
B-9	905	25	< 0.05	0.05
B-9	906	30	< 0.05	0.05
B-10	1000	2	< 0.05	0.05
B-10	1001	5	0.09	0.05
B-10	1002	10	< 0.05	0.05
B-10	1003	15	< 0.05	0.05
B-11	1100	2	< 0.05	0.05
B-11	1101	5	< 0.05	0.05
B-11	1102	10	< 0.05	0.05
B-12	1200	2	< 0.05	0.05
B-12	1201	5	< 0.05	0.05
B-12	1202	10	< 0.05	0.05

FIELD BLANKS				
Sample No.	2-Hexanone (ug/L)	Detection Limit (ug/L)		
1108	<50	<50		
1400	<50	<50		